# **Explosion of Lycopodium Powder**

## **Purpose**

To demonstrate the effect of surface area on the rate of a chemical reaction.

#### **Materials**

exploding can device lycopodium power spatula/spoon 100 mL beaker lighter

#### **Procedure**

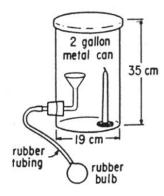
- 1. Place lycopodium powder in the device.
- 2. Carefully light the birthday candle in the device.
- 3. Squeeze firmly on the pipette bulb, and watch as the dust cloud from the lycopodium lights.
- 4. After the flame cloud is produced, use the beaker to snuff the candle. DO NOT BLOW OUT CANDLE, this can cause the lycopodium powder to ignite and burn your face.



#### **Additional Information**

- 1. This demonstration illustrates that a combustible material, as lycopodium powder, when finely divided and dispersed in the air, will explode upon ignition.
- 2. The powder will not ignite or burn, however, unless it is air-borne as dust.
- 3. Any spilled powder can be wiped up with a damp cloth and discarded in a waste container.
- 4. Tossing very small amounts of lycopodium powder into an open flame will have sparkling results.
- 5. Variation:
  - a. Assemble the explosion can as diagramed below:

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- b. Be sure the candle wick is level with the funnel top.
- c. With a spatula, add a small amount of lycopodium powder to the funnel.
- d. Put the shield in place and turn off the lights.
- e. Light the candle. Tongs will be useful. Fireplace matches also work well.
- f. Seal the container uniformly (if the lid is to fly straight up).
- g. Quickly move as far away as possible, while holding the rubber bulb, and squeeze the bulb firmly.

### Reference

Aylea, H.N. & Dutton, F.B., Tested Demonstrations in Chemistry, 1965.

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